

## Indo-American College

Not just another college

Permanently Affiliated to THIRUVALLUVAR UNIVERSITY, Vellore. Accredited by NAAC with 'B' Grade Recognised Under Section 2 (f) & 12 (b) of UGC Act.

T.N.Govt.G.O.MS. No.172, Higher Education Dept, dt.27-4-1998

## **DEPARMENT OF CHEMISTRY**

## PROGRAMME OUTCOMES (B. Sc)

PO1:	Describe the basic concepts, fundamental principles, and the scientific theories related to various scientific phenomena and their relevancies in day-to-day life
PO 2:	Employ critical thinking for solving problems using basic chemistry knowledge and concepts.
PO 3:	Acquire skills in handling scientific instruments, planning and performing laboratory experiments and drawing logical inferences from the chemical experiments.
PO 4:	Analyze the given scientific data critically and systematically to draw a logical conclusion.
PO 5:	Develop various communication skills such as reading, listening, speaking, etc., to express ideas and views clearly and effectively.
PO 6:	Create an intellectual curiosity and ability to think in a scientific manner and get sensitized to social and environmental realities.
PO 7:	Develop an interest in pursuing higher studies in Chemistry and related subjects which are relevant to employment and entrepreneurship.
PO 8:	Capable of self-paced and self-directed learning aimed atpersonal developmentand for improving knowledge/skill development and reskilling.
PO 9:	Integrate the knowledge and skills developed in multidisciplinary environments and function effectively as an individual or a leader and contribute towards the needs of the society.

## **COURSE OUTCOME**

S. No.	Course Code	Course Title	Course Outcomes
1	FCH11	General Chemistry-I	<ul> <li>At the end of the course, students should be able to:</li> <li>LO1. Understand about atomic structure, able to write electronic configuration</li> </ul>

			<ul> <li>and knowing periodic properties and classification of elements.</li> <li>LO2. Able to calculate the bond-order, to MO diagram of molecules. Understand about types of bonds.</li> <li>LO3. Acquired knowledge about basic concepts of organic chemistry, able to written the IUPAC name of organic compounds. Get knowledge about inductive effect.</li> <li>LO4.students should be able to describe the characteristics of the three states of matter and types of crystals.</li> <li>LO5. Got knowledge about volumetric quantitative analysis and its calculation.</li> </ul>
2	FPE10C	Professional English-I	<ul> <li>At the end of the course, students should be able to:</li> <li>LO1. Develop the communication skills through the multiple tasks. Develop the English spoken skills.</li> <li>LO2. Develop the writing skills about any topic and story writing.</li> <li>LO3. Negotiation Strategy skills developed.</li> <li>LO4. Develop the presentation skill</li> <li>LO5. Develop the critical thinking skill.</li> </ul>
3	FCH21	General Chemistry-II	<ul> <li>At the end of the course, students should be able to:</li> <li>LO1. Understand the groups of s and p block elements and its chemical and physical properties.</li> <li>LO2. Derive the reaction mechanism of hydrocarbons and organic compound preparation.</li> <li>LO3. Acquired knowledge about Dines of organic compounds.</li> <li>LO4. Understand about quantum chemistry and thermo chemistry</li> <li>LO5. Recognize the basic terms of thermo dynamics and able to predict the energy changes.</li> </ul>

4	FPE20C	Professional English- II	<ul> <li>At the end of the course, students should be able to:</li> <li>LO1. To improve the communication competence.</li> <li>LO2. Develop the persuasive communication.</li> <li>LO3. Enhance the digital competence.</li> <li>LO4. Improve the creativity and imagination.</li> </ul>
			LO5.Develop the workplace communication and the basics of academic writing.
5	ССН31	General Chemistry-III	<ul> <li>At the end of the course, students should be able to:</li> <li>LO1. Understand the inorganic qualitative analysis and spot test reagents, types of solvents.</li> <li>LO2. Understand the comparative study of Carbon, Nitrogen and Oxygen family elements.</li> <li>LO3.Got knowledge about mechanism for electrophilic substitution and aromaticity.</li> <li>LO4.Recognize the mechanism for aliphatic nucleophilic substitution, elimination, aromatic nucleophilic substitution reaction.</li> <li>LO5. Recognize the basic concept of second law of thermodynamics and entropy.</li> </ul>
6	CCH41	General Chemistry- IV	<ul> <li>At the end of the course, students should be able to:</li> <li>LO1. The students will be able to state the resemblances of elements within each main group in noble gases and clathrates.</li> <li>LO2. Understanding the types of carboxylic acids and amines.</li> <li>LO3. Study about alcohol, phenols and naphthols.</li> <li>LO4. Known about the basic concepts of thermodynamics.</li> <li>LO5. Able to understand the physical significance of third law of thermodynamics and entropy.</li> </ul>

7	CPCH22	Practical –I Volumeteric Analysis	<ul> <li>At the end of the course, students should be able to:</li> <li>LO1.Facilitate the learner to make solutions of various molar concentrations.</li> <li>LO2.The concept of mole: converting moles into gram; converting gram into moles.</li> <li>LO3.Defining concentration; dilution of solutions; making different molar concentrations.</li> <li>LO4.Acquired knowledge about volumetric quantitative analysis experimentally.</li> <li>LO5. They were able to calculate the amount of substances from acidimetry, dichromometry, idometry, complexometry and precipitation titration.</li> </ul>
8	FES10	Environmental Studies	<ul> <li>At the end of the course, students should be able to:</li> <li>LO1. Understand the Green house effect, soil, water and air pollution, acid rain, etc.</li> <li>LO2. Apply the knowledge to aware common people about environmental pollution.</li> <li>LO3. Do more research on waste management, nuclear waste management, biodegradation of hazardous wastes etc.</li> <li>LO4. How to protect the forest</li> <li>LO5.students got the awareness about social act and rules.</li> </ul>
9	FGA20	Value Education	<ul> <li>At the end of the course, students should be able to:</li> <li>LO1. Students were got knowledge about value of education.</li> <li>LO2. Students got idea about how to lead the family in society.</li> <li>LO3.able to personality development.</li> <li>LO4. Understood about social awareness and consumer rights.</li> <li>LO5. Students got idea about modem warfare and terrorism.</li> </ul>

10	СРСН 44	Core Practical - II	<ul> <li>At the end of the course, students should be able to:</li> <li>LO1. Able to analyze ions present in the inorganic mixture.</li> <li>LO2. Able to carry out scientific semi micro qualitative analysis</li> <li>LO3. Acquired knowledge about the inorganic chemical reactions</li> <li>LO4. Able to prepare to inorganic compounds</li> <li>LO5. Learned about handling of various chemicals and reagents.</li> </ul>
11	CSCH 34	Water Treatment And Analysis	<ul> <li>At the end of the course, students should be able to:</li> <li>LO1: Students well known about water and its characteristics; hard water; purification of hard water; and students got knowledge about sterilisation and disinfection of water.</li> <li>LO2: students got idea about water softening methods like Clarks, lime soda, zeolite and deminerlisation; and determination of harness of water by titration method; how to calculate temporary and permanent hardness.</li> <li>LO3: students got exposure of industrial water treatment methods; desalination, electrolysis, reverse osmosis; effluent treatment of water from paper industry, petrochemicals, fertilizers and power station.</li> <li>LO4: acquire knowledge about water ananlysis and pollution of water from fertilizers, detergents and pesticides industries.</li> <li>LO5: student got knowledge about analysis of chemical substance present in the water samples.</li> </ul>
12	CSCH44	Food Chemistry	<ul> <li>At the end of the course, students should be able to:</li> <li>LO1. Students able to understand about cereals and its nutritive values; sugar and sugar related products and its advantages and disadvantages.</li> <li>LO2. Students well known about</li> </ul>

			<ul> <li>vegetables, fruits, classification, composition and its nutritive values.</li> <li>LO3. Knowing the value of beverages types of beverages and its nutritive values.</li> <li>LO4. Got idea about food preservation and its methods.</li> <li>LO5. Acquired knowledge about food additives, food packing and food colouring.</li> </ul>
13	CCH52	Inorganic Chemistry-I	<ul> <li>At the end of the course, students should be able to:</li> <li>LO1. Students get understood the properties of halogens and their compounds</li> <li>LO2. Acquired knowledge about coordination compounds and their applications and able to named coordination compounds</li> <li>LO3.Students well understood theories of co-ordinations compounds and able to calculate the stabilization energies.</li> <li>LO4.Knowing the applications about coordination compounds in qualitative and quantitative analysis.</li> <li>LO5. Students get knowledge about solid state and applications.</li> </ul>
14	CCH62	Inorganic Chemistry- II	<ul> <li>At the end of the course, students should be able to:</li> <li>LO1. Acquired knowledge about nuclear chemistry and applications of radioisotopes.</li> <li>LO2. Applying the knowledge of radioactivity in nuclear reactors and applications of nuclear reactions.</li> <li>LO3.Understood about metallurgy process of metals from ores.</li> <li>LO4. Learn about the inner transition elements.</li> <li>LO5.Get knowledge about Organometallic compounds and their applications.</li> </ul>
15	СРСН66	Core Practical –III	<ul> <li>At the end of the course, students should be able to:</li> <li>LO1. Applying the knowledge of</li> </ul>

			<ul> <li>gravimetric quantitative estimations.</li> <li>LO2. Students get idea about estimation of sulphate as barium sulphate using silica crucible.</li> <li>LO3. Able to estimate the amount of solute present in the precipitate.</li> <li>LO4. Able to estimate the barium as barium chromate using sintered crucible with accuracy</li> <li>LO5.Able to carry out scientific experiments as well as accurately record and analyse the result.</li> </ul>
16	CCH52	Organic Chemistry-I	<ul> <li>At the end of the course, students should be able to:</li> <li>LO1. In order to study carbohydrates will develop the skills to recognize and draw particular carbohydrate structures.</li> <li>To know general structural elements of cyclic monosaccharide and disaccharides and their implications for structure and function.</li> <li>LO2. To be able to recognize the types of isomerism.</li> <li>LO3. The reactivity and stability of an organic molecules based on structure, including conformation and stereochemistry.</li> <li>LO4. The prediction of organic reaction and mechanisms.</li> <li>LO5. To develop novel, efficient, convenient, selective and environmentally benign synthetic methods in organic chemistry.</li> </ul>
17	ССН62	Organic Chemistry-II	<ul> <li>At the end of the course, students should be able to:</li> <li>LO1.Able to know about the mechanism involved in molecular rearrangements in organic reaction and some naming rearrangements.</li> <li>LO2. Learn about Aminoacids their types and properties, structure of pepties and their synthesis.</li> <li>LO3.Understand about proteins and their classification, structure, types of nucleic acids and their constituents.</li> </ul>

			<ul> <li>LO4. Learn about antibiotics and their therapeutic activity, Identification of alkaloids and terpenoids from plants and their structural elucidation and activity.</li> <li>LO5. Able to understand about organo synthetic reagents and their preparation and synthetic application for large scale</li> <li>Synthesis in industry.</li> </ul>
18	СРСН67	Core Practical –IV Organic Qualitative Analysis And Preparations.	<ul> <li>At the end of the course, students should be able to:</li> <li>LO1.Identification of organic compounds with one functional group like aldehydes, ketones, ester, phenol, anilide and nitrocompounds.</li> <li>LO2. Able to identify mono and dicarboxylic acids, reducing and non reducing sugars, mono and diamides.</li> <li>LO3.Demostrate some of the organic compounds by organic preparations.</li> <li>LO4.Finally knows about how to handling the chemicals carefully.</li> <li>LO5. Able to know laboratory practices and safety.</li> </ul>
19	CCH53	Physical Chemistry –I	<ul> <li>At the end of the course, students should be able to:</li> <li>LO1.Able to understand the solutions liquid to liquids and Nernst law and its application.</li> <li>LO2.students able to explain about phase rule terms and definitions; this may included able to draw phase diagram of one and two components system.</li> <li>LO3.students understood about osmotic pressure, vant hoff factor and chemical equilibrium.</li> <li>LO4.State the basic principles electrochemistry. Mention and explain various methods for the determination of transport number.</li> <li>LO5. To understand application of conductometric titrations and concept of PH.</li> </ul>
20	ССН63	Physical Chemistry – II	• At the end of the course, students should be able to:

			<ul> <li>LO1. Students able to understand cell reactions and emf.</li> <li>LO2. Gain knowledge about fuel cells and storage cells.</li> <li>LO3. Students Derive the integrated rate expressions for zero order, first order, second order and third order reaction.</li> </ul>
			<ul> <li>Understand theories of reaction kinetics and differentiate them.</li> <li>L04. Students gain the knowledge about adsorption properties and activity of catalyst.</li> <li>LO5. Clearly understood about photochemistry and its applications.</li> </ul>
21	СРСН68	Core Practical–V Physical Chemistry Experiments	<ul> <li>At the end of the course, students should be able to:</li> <li>LO1. Explain the principle behind the experiments performed in the laboratory.</li> <li>LO2. Plan and perform experiments and interpret experimental results.</li> <li>LO3. Students got idea about various heating reactions via phase rule.</li> <li>LO4. Students able understand the kinetics with respect to time.</li> <li>LO5.students understanding about electrolytes with related experiments.</li> </ul>
22	CECH54 A	Analytical Chemistry- I	<ul> <li>At the end of the course, students should be able to:</li> <li>LO1. Able to minimize the laboratory calculation error.</li> <li>LO2. Students got idea about purification compounds, gravimetric analysis.</li> <li>LO3. Able to understand about basic principles and applications of microwave and UV spectroscopy.</li> <li>LO4. Able to understand about basic principles and applications of IR spectroscopy.</li> <li>LO5. Able to understand about basic principles and applications of Raman spectroscopy.</li> </ul>
23	CSCH64A	Analytical Chemistry- II	At the end of the course, students should be able to:

			<ul> <li>LO1. Able to understand about basic principles and applications of chromatography.</li> <li>LO2. Able to understand about basic principles and application of polarography.</li> <li>LO3. Able to understand about basic principles and applications of NMR spectroscopy.</li> <li>LO4. Able to understand about basic principles and applications of MASS and ESR spectroscopy.</li> <li>LO5. Able to understand about basic principles and application of TGA and DTA.</li> </ul>
24	CECH54 A	Pharmaceutical Chemistry	<ul> <li>At the end of the course, students should be able to:</li> <li>LO1. Students understanding of the basic pharmacological terms.</li> <li>LO2. Able to diagnostic tests of diseases and disorder.</li> <li>LO3 get knowledge about anti microbial drugs.</li> <li>LO4 Acquired knowledge about anesthetics and analgesics.</li> <li>LO5. Got knowledge about hormones and its physiological function</li> </ul>
25	CSCH56	Applied Chemistry	<ul> <li>At the end of the course, students should be able to:</li> <li>LO1. Students can apply the knowledge in petro chemical industries.</li> <li>LO2. Students get idea about paper manufacturing industries.</li> <li>LO3. Acquired knowledge about sugar industries.</li> <li>LO4. Understood about explosive compounds</li> <li>LO5. Got knowledge about dairy industries and products.</li> </ul>
26	CSCH66	Agricultural And Leather Chemistry	<ul> <li>At the end of the course, students should be able to:</li> <li>LO1. Students acquired basic knowledge of properties of soil and soil fertility</li> <li>LO2. Accumulated skill for scientific research work in agricultural field.</li> </ul>

		<ul> <li>LO3. Students able to understand action of pesticides.</li> <li>LO4. Students got exposure about leather technology.</li> <li>LO5. Students acquired knowledge about process tanning effluents treatments</li> </ul>
27	Group Project	<ul> <li>Get knowledge about research and its methodology, To got idea about synthesis of compounds, Students able to understand characterization methods, Got idea about</li> <li>Viva- voice.</li> </ul>